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Zn(OH)₂ + NO₂. Identify the species being oxidized and reduced in each of the following reactions:

a. $\text{Cr} + \text{Sn}^{4+} \rightarrow \text{Cr}^{3+} + \text{Sn}^{2+}$

b. $3 \text{Hg}^{2+} + 2 \text{Fe}(\text{s}) \rightarrow 3 \text{Hg}(\text{s}) + 2 \text{Fe}^{3+}$

c. $2 \text{As}(\text{s}) + 3 \text{Cl}_2(\text{g}) \rightarrow 2 \text{AsCl}_3(\text{s})$

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