

Modern Chemistry Chapter 2 Test

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CHAPTER 2 REVIEW Measurements and Calculations SECTION 1 SHORT ANSWER Answer the following questions in the space provided. 1. Determine whether each of the following is an example of observation and data, a theory, a hypothesis, a control, or a model. observation and data a. A research team records the rainfall in inches

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Modern Chemistry 105 Chapter Test Name Class Date Chapter Test A, continued Use this figure to answer questions 7 and 8. _____ 7. A solution containing 35 g of Li₂SO₄ dissolved in 100 g of water is heated from 10 ° C to 90 ° C. According to information in the figure, this temperature change would result in a. an additional 5 g of Li₂SO₄ in ...

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Modern Chemistry 86 Chapter Test Name Class Date Chapter Test A, continued _____ 13. The separation process of paper chromatography can be explained by a. vaporization of the ink. b. water vapor pressure. c. capillary action. d. pull of gravity. _____ 14. When there is a small decrease in temperature, the average kinetic energy of the particles ...

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18. Dalton's atomic theory acyreed with the modern atomic theory EXCEPT for the statement that all atoms of the same element have the same Complete the following table to compare the types of subatomic particles. 19. 20. 21. 10 Particle Proton Neutron Electron CHAPTER 3 TEST Mass n mber Relative charge Locatio MODERN CHEMISTRY

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CHAPTER 6 REVIEW Chemical Bonding SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. a The notation for sodium chloride, NaCl, stands for one (a) formula unit. (c) crystal. (b) molecule. (d) atom. 2. d In a crystal of an ionic compound, each cation is surrounded by a number of (a) molecules. (c) dipoles. (b) ...

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